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Energy is the capacity to produce change. It is a fundamental property of any system, and it should = , : $w = \int \mathbf{w} = \mathbf{M}$, $= \int \int = \int \int$. 2

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Thermodynamic Concepts. 2.1 The Thermodynamic Perspective. e

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defined geochemistry as the application of chemical knowledge and techniques to solve geo- logical problems.

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W. M White Geochemistry Chapter 11: The Mantle and Core 487
 $d\rho(r) dr = G r^2 \rho(r) V P^2 \rho^4 3 V s^2 \rho^4 \rho(r)r^2 dr$ 11.12 Equation
11.12 describes how density changes in a self-compressing, but
otherwise uniform sphere and is known as the Adams-Williamson

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Equation. m \rightarrow PREM

W. M White Geochemistry Chapter 11: The Mantle and Core

W. M. White Geochemistry Chapter 7: Trace Elements November 21, 2007263 typically 10^{-4} to 10^{-12} STP cm³/g (10^{-1} to 10^{-9} ppm).

Their solubility in silicate melts is a strong function of pressure, as well as both atomic radius and melt composition as is illustrated in Figure 7.4.

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W. M. White Geochemistry Chapter 5: Kinetics © W. M. White 2011 158 5.2.3 Reaction Rates Consider a reaction such as the

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precipitation of dolomite from a solution.

W. M. White Geochemistry Chapter 5: Kinetics C 5: K T P T (4.5 / 5.0 \square 3 customer ratings) This book provides a comprehensive introduction to the field of geochemistry. The book first lays out the \square geochemicaltoolbox \square : the basic principles and techniques of modern geochemistry, beginning with a review of thermodynamics and kinetics as they apply to the Earth and its environs.

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William White teaches geochemistry as a Professor of earth and atmospheric sciences at Cornell University. He received a B.A. in

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Geochemistry 320 January 10, 2001 also binds quarks together to form hadrons, a class of particles that includes neutrons and protons. The intensity of the strong force decreases rapidly with distance, so that at distances more than about 10-14 m it is weaker than the electromagnetic force.

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in this chapter we will consider the behavior of trace elements,

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particularly in magmas, and introduce methods to model this behavior. Though trace elements, by definition, constitute only a small fraction of a system of interest, they provide geochemical and geological information out of proportion to their abundance.

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W. M. White Geochemistry Chapter 10: Cosmochemistry

418 July 31, 2006 we learn about the evolution of the Earth by examining old rocks, we can learn about the evolution of the cosmos by looking at old stars. The old stars of Population II are considerably poorer in heavy elements than are young stars.

W. M. White Geochemistry Chapter 10: Cosmochemistry ...

W. M. White Geochemistry Chapter 4: Applications of

Thermodynamics 120 October 17, 2001 $m = m + + 22 \ 2 \ o \ RT \ X \ W$

In 4.15 G Equation 4.14 is Raoult's Law; letting: $\mu^* = \mu^\circ + W G$ or

$W G = RT \ln h$ then 4.15 is Henry's Law. Thus the interaction

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parameter can be related to the parameters of Henry's Law, and activity coefficient. In the Mar-

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Geochemistry Chapter 3: Solutions William White teaches geochemistry as a Professor of earth and atmospheric sciences at

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W M White Geochemistry Chapter W. M. White Geochemistry Chapter 7: Trace Elements 261 HCO_3^- , Mg^{2+} , Ca^{2+} , K^+ and Na^+ (and H_2O , of course) can be considered a trace constituent, though Sr^{2+} , HBO_3^- , and Br^- are sometimes considered major constituents also (constituents or species is a better term here than elements).

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W. M. White Geochemistry Chapter 7: Trace Elements

W. M. White Chapter 9: Stable Isotopes. Geochemistry 9.2.1.1 The Quantum Mechanical Origin of Isotopic Fractionations. It is fairly easy to understand, at a qualitative level at least, how some isotope

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fractionations can arise from vibrational motion.

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