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Theory Name

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, Gas molecules

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the molecules is

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temperature of  
the gas. Ideal  
Gas Gas whose  
conforms to the  
Boyle's Law - Pressure  
area.

Atmospheric

Pressure -

Barometer: Gas

Pressure:

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Kinetic Theory,

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Theory • Gases are composed of small, separate particles called molecules. • Gas molecules are in constant motion. • All collisions between particles are perfectly elastic. • The molecules of a gas display no

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attraction or  
repulsion for  
one another.

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