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Quantum Numbers, Atomic Orbitals, and

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Chapter 8 - Quantities

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Chapter 7 Fundamental

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Modern Chemistry 71

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B, continued 17. A

substance combines

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with oxygen, releasing a large amount of energy as heat and light, in a(n)

. 18. The decomposition of a substance by an electric current is called

. 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

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Test B~~

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Modern Chemistry 141

Chapter Test Name

Class Date Chapter Test

B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is

than 7. 19. A is a highly purified solid used to check the concentration of a standard solution.

20. A 1 M solution of

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NaOH will have a pH
that is

Chapter Test

With Answer
Assessment Chapter

Test B — Baumapedia

Modern Chemistry 137

Chapter Test Name

Class Date Chapter Test

A, continued _____ 19.

In the figure on the
previous page, the pH at
the equivalence point a.
is equal to 7.0. b. is
greater than 7.0. c. is

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less than 7.0. d. cannot
be determined from the
data given. _____20. In
the figure on the
previous page, the
volume of titration
standard

~~Assessment Chapter~~
~~Test A~~ ~~Baumapedia~~
Modern Chemistry 105
Chapter Test Name
Class Date Chapter Test
A, continued Use this

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Chemistry

figure to answer
questions 7 and 8.

Chapter Test
_____ 7. A solution

With Answer
containing 35 g of Li

2SO₄ dissolved in 100

g of water is heated

from 10°C to 90°C.

According to

information in the

figure, this temperature

change would result in

a. an additional 5 g of Li

2SO₄ in ...

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18. Dalton's atomic theory agreed with the modern atomic theory

EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19.

20. 21. 10 Particle

Proton Neutron Electron

CHAPTER 3 TEST

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Mass number Relative
charge Locatio

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Chapter 13: Ions in

Aqueous Solutions and

Colligative Properties

Chapter Practice Test

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Chapter 7 Test Modern

Chemistry Chapter 7

test. STUDY. PLAY.

The basis of an ionic bond is the__ absorption of water into their solid structures. Elements tend to gain or lose electrons in order to acquire the electron configuration of a__ noble gas. Ions that are formed from more than

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one atom are called__

Chapter Test

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Chapter 7 Test B

Answers

CH 1 Reading

Assignment Modern

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Vocabulary-New. CH 1

Mixed Questions. CH 1

Matter & Energy

Vocabulary-New. CH 1

Classifying Matter-

New. CH 1

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Chemistry of Matter

... CH 14 Chapter Test .

CH 14 Acid Base Basics

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14 Practice w answers.

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16.1-16.3. White Board

Practice: CH 14 Naming

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Answer Key -

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Capillary Action. Fluid.

Surface Tension.

Vaporization. the

attraction of the surface

of a liquid to the surface

of a.... a substance that

can flow and therefore

take the shape of its.... a

force that tends to pull

adjacent parts of a

liquid's surfac....

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Wikispaces the answer

must not have more

significant figures than

the . number in the

calculation with the

fewest significant [PDF]

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