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the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g
and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed
with 12.0 mol of H₂ according to the following equation: N₂(g) + 3H₂(g) → 2NH₃(g) N

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following questions in the space provided. 1. a.

6 Chemical Bonding

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