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Match description on the right to the correct crystal type on the left  
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covalent molecular crystal (b) is hard, brittle, and nonconducting a  
metallic crystal (c) typically has the lowest melting point of the four

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW  
Chemical Bonding SECTION 4 SHORT ANSWER Answer the  
following questions in the space provided. 1. \_\_\_\_\_ In metals, the  
valence electrons are considered to be (a) attached to particular  
positive ions.

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