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of electrons between the oxygen and hydrogen atoms. 9. The concentration of H ions determines whether a solution is acidic or basic.

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CHAPTER 2 REVIEW

Measurements and Calculations

SECTION 1 SHORT ANSWER

Answer the following questions in
the space provided. 1. Determine
whether each of the following is
an example of observation and
data, a theory, a hypothesis, a
control, or a model. observation
and data a. A research team
records the rainfall in inches

2 Measurements and Calculations

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Test A, continued Use this figure
to answer questions 7 and 8.

_____ 7. A solution containing 35 g
of Li_2SO_4 dissolved in 100 g of
water is heated from 10°C to
 90°C . According to information in
the figure, this temperature
change would result in a. an
additional 5 g of Li_2SO_4 in ...

Assessment Chapter Test A - Ed
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20 shows that 2 molecules (made
of 4 atoms) of hydrogen and 1
molecule (made of 2 atoms) of
oxygen produce 2 molecules of
water. The total mass of the
product, water, is equal to the
sum of the masses of each of the
reactants, hydrogen and oxygen.

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What parts of Dalton's atomic theory are illustrated by this reaction?

3 Atoms: The Building Blocks of Matter

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PROBLEMS Write the answers to the questions on the line to the left, and show your work in the space provided. 25. 26. MODERN

CHEMISTRY The atomic number of nickel-60 is 28. How many neutrons does this isotope have? Carbon-14 has 8 neutrons.

What is the atomic number of carbon-14? 11 CHAPTER 3 TEST

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CHAPTER 1 REVIEW Matter and
Change MIXED REVIEW SHORT

ANSWER Answer the following
questions in the space provided.

1. Classify each of the following
as a homogeneous or
heterogeneous substance.

homogeneous a. sugar

homogeneous d. plastic wrap

homogeneous b. iron filings

heterogeneous e. cement

sidewalk heterogeneous c.

granola bar 2. For each type of

investigation, select the most
appropriate ...

1 Matter and Change -

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CHAPTER 10 REVIEW States of
Matter SECTION 3 SHORT
ANSWER Answer the following
questions in the space provided.
1. Match description on the right

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to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms.
Agostine's Chemistry Page
Algebra 1: Common Core (15th Edition) Charles, Randall I.
Publisher Prentice Hall ISBN
978-0-13328-114-9

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Chemistry CHAPTER 22 REVIEW
Organic Chemistry SECTION 1

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Chapter 5
SHORT ANSWER Answer the following questions in the space provided. 1. Name two types of carbon-containing molecules that are not organic. _____ 2. _____
Carbon atoms form bonds readily with atoms of ...