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of the following binary compounds of non-metallic elements a. CBr₄ b. N₂P₃ c. PCI₃ d. ICl e.

N₂O f. SiF₄ 2. Write the name for the following compounds of nonmetallic elements. a. GeH₄

b. N₂Br₄ c. P₂S₅ d. SeO₂ e. NH₃ f. SiO₂ 3.

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5.1.1 Exercise 1 - concentration-time graphs 5.1.1 Exercise 2 - orders of reaction 5.1.2 Exercise 1 - calculating equilibrium constants 5.1.2 Exercise 2 - equilibrium constants and the equilibrium position

~~Unit 5—A-Level Chemistry~~

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1. Name _____ Chemistry – Unit . 5. Worksheet . 2. An old (pre-1987) penny is nearly pure copper. If such a penny has a mass of 3.3 g, how many moles of copper atoms would be in one penny?

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Any other units can be constructed as a combination of fundamental units. For example, velocity could be measured in meters per second (written m/s or $m \times s^{-1}$), and area could be measured in units of meters squared (m^2). When a named unit is defined as a combination of base units, it is called a derived unit. For example, the SI unit of energy is the joule (J), which is defined as a $kg \times m \dots$

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