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PLAY. What is a chemistry mole? SI unit for a quantity. What is molar mass? Mass of one mole of a substance. What is Avogadro's number? 6.02×10^{23} particles. What is stoichiometry? Quantitative study of reactants used up and product amounts made. Upon what chemistry law is stoichiometry based? Law of conservation of matter/mass. What ...

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discoveries ...

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solution. A reaction that ABSORBS energy in the form of heat. A reaction that releases energy in the form of heat. A solution that contains the maximum amount of dissolved solute. Any solution that can dissolve more solute at a given temperat... endothermic reaction. A ...

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EXAM 2 (see subsections ... equipment are always present in the
Chemistry Building. Final Exam: The final (200 points) is all
-inclusive. University rules stipulate that a course grade of 0.0 is
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questions below. Turn this in on the day of your final. • Read
BELOW!!!! • All answers must be ... a. hand written, b. numbered,
c. on a separate sheet of paper,

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