

Chemical Principles Zumdahl Solutions Manual

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chemical reaction always equals the total mass after a chemical reaction. Law of definite proportion: a given compound always contains exactly the same proportion of elements by mass. For example, water is always 1 g hydrogen for every 8 g oxygen. Law of multiple proportions: When two elements form a series of compounds, the ratios of

CHAPTER 2 ATOMS, MOLECULES, AND IONS

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