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Chemistry: Supplemental Stoichiometry Problems. Directions: Solve each of the following problems. Assume excess of any reactant that isn't mentioned, unless otherwise specified. Show your work, including proper units, to earn full credit. 1. Given the balanced equation, show what the following molar ratios (i.e., the conversion factors) should be.

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Supplemental Problems - MARRIC

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Chapter 12 Stoichiometry Answer Chapter 12 Stoichiometry Packet Answer Key Answer:  $4.93 \times 10^{-5}$  L or 49.3  $\mu$  L In Example 12.2.1 and Example 12.2.2, the identity of the limiting reactant has been apparent:  $[\text{Au}(\text{CN})_2]^-$ ,  $\text{LaCl}_3$ , ethanol, and para-nitrophenol. Stoichiometry Packet Answers Chapter 12 Chapter 12 Stoichiometry Packet Answers

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