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*Finding the
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ions for a mixed
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_____ Complete the

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following problems
on a separate
sheet of paper. Use
significant figures.

Note: The density
of water is 1 g/mL.

1. What is the
molarity of a
solution that
contains 10.0
grams of Silver
Nitrate that has
been dissolved in
750 mL of water?

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Answers 1) If I

make a solution by
adding 83 grams of
sodium hydroxide
to 750 mL of
water... To solve
problem 1, you
need to have
calculated for
various parts that
there are 2.08

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moles of NaOH
(which has a molar
mass of 40 g/mol),
that there are 750
grams of water
(which has a
density of 1 g/mL),
and that there ...

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Percent-by-volume
concentration of a

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of solution. Calculate the volume percent, $\%(v/v)$, of solute in the following solution: 20.0 mL of methyl alcohol in enough water to give 475 mL of solution.

Mass-volume percent concentration.
Normal saline solution that is

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used to dissolve
drugs for
intravenous use is
0.92%(m/v) NaCl in
water.